Homework #6; Due Monday 3/16. (Review on 3/16; Midterm 2 on 3/18.)

NOTE: Most, if not all, of the integrals you need to do this homework are in Appendix 3 or 6. Or you can just look them up...

- 1. Explicitly show that the sum of **any** two solutions to the Schrodinger equation will lead to a third solution. (Don't assume that the solutions take any particular form.)
- 2. Normalize the stationary states of the infinite square well. Do it for arbitrary "n".
- 3. A particle with mass M is in the second-lowest energy state in an infinite square well of width 'a'. What is the probability it will be found within a distance D from one of the two walls? (Assume D < a/2. Check your answer for the special case D=a/2.)
- 4. Normalize the wavefunction $\psi(x) = A \exp\left(-\frac{x^2}{a^2}\right)$. (See Appendix 6.) "a" is known;

Here A is the normalization constant. This wavefunction is not in a box; it goes to infinity.

- 5. (4 points). At t=0, an electron in a one-dimensional box of length L is a superposition of the lowest two energy states; $\psi = 3\psi_1 + 4\psi_2$. You can assume that ψ_1 and ψ_2 are **individually** normalized, but ψ is (obviously!) not normalized.
- A) Normalize the full wavefunction without plugging in for ψ_1 or ψ_2 . (There are two ways to do this!)
- B) What does the explicit wavefunction look like as a function of (x,t) at a later time?
- C) Where is the most likely place that you will find the particle at a given time t? (Where is the probability maximum?) Make sure you plug your full answer from part B) so that your answer is a function of time.
- D) Show that your answer to C) is real, and describe its behavior with time.
- E) What might you find if you made an energy measurement of the electron, and with what probabilities? (You can do this at t=0, because the answer doesn't change with time.)
- F) Suppose you measured the maximum possible value of the energy, that it is possible to measure given this particular wavefunction. What would the "collapsed" wavefunction look like?